

An Efficient Triple Allylation of 1,3,5-Triazine by Means of Tin Reagent.  
Facile Synthesis of syn,syn-2,4,6-Trifunctionalized 1,3,5-Triazacyclohexanes

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An efficient triple addition of allyl groups to 1,3,5-triazine has been carried out by means of an allyltin reagent to give 2,4,6-triallyl-1,3,5-triazacyclohexanes in high yields. Epoxidation and hydroxylation of the syn,syn-stereoisomer afford geometrically interesting syn,syn-trifunctionalized 1,3,5-triazacyclohexane derivatives.

Development of efficient allylation reactions using allylic metal reagents have been one of the key objectives in synthetic organic chemistry.<sup>1)</sup> We have uncovered that allylic tin reagents react with nitrogen heteroaromatics and cyclic imines activated by a variety of acyl chlorides chemo- and/or regioselectively.<sup>2)</sup> The important point is that tin reagents do not react either with acyl chlorides or with C=N bonds, but they do react with *N*-acylated iminium ions. In light of the high chemoselectivity, we have anticipated that a polyazaaromatic compound may be polyallylated with stoichiometric amounts of an allyltin reagent in the presence of stoichiometric amounts of an acyl chloride. We report herein that 1,3,5-triazine reacts chemoselectively with 3 equivalents of allyltributyltin in the presence of 3 equivalents of chloroformate esters to give 2,4,6-triallyl-1,3,5-triazacyclohexanes very efficiently.

When 3 equivalents of methyl chloroformate were added to a mixture of 1,3,5-triazine (**1**) and 3 equivalents of allyltributyltin (**2**), a triple addition reaction readily proceeded to give syn,syn- and syn,anti-2,3,4-triallyl-1,3,5-trimethoxycarbonyl-1,3,5-triazacyclohexane (**3a** and **3b**) in a ratio of 20:80 in 87% yield. The stereochemistry of **3a** and **3b** was deduced by <sup>1</sup>H and <sup>13</sup>C NMR: the NMR spectra of **3a** exhibit higher symmetry than those of **3b**.<sup>3,4)</sup>

It has been found that the ratio of the stereoisomer is dependent on the kind of chloroformate ester. The results are summarized in the Table 1. When 2,2,2-trichloroethyl chloroformate was used as activating agent, the ratio of syn,syn-:

syn,anti-stereoisomers (**4a** : **4b**) was 40:60. Furthermore, the reaction of **1** with **2** in the presence of phenyl chloroformate gave syn,syn- and syn,anti-stereoisomer (**5a** and **5b**) in a ratio of 42:58 in 92% yield. In a statistical point of view, the percentage of the syn,syn-stereoisomer should be 25%. Thus, sterically demanding chloroformate esters might increase the ratio of syn,syn-stereoisomer.<sup>5)</sup>

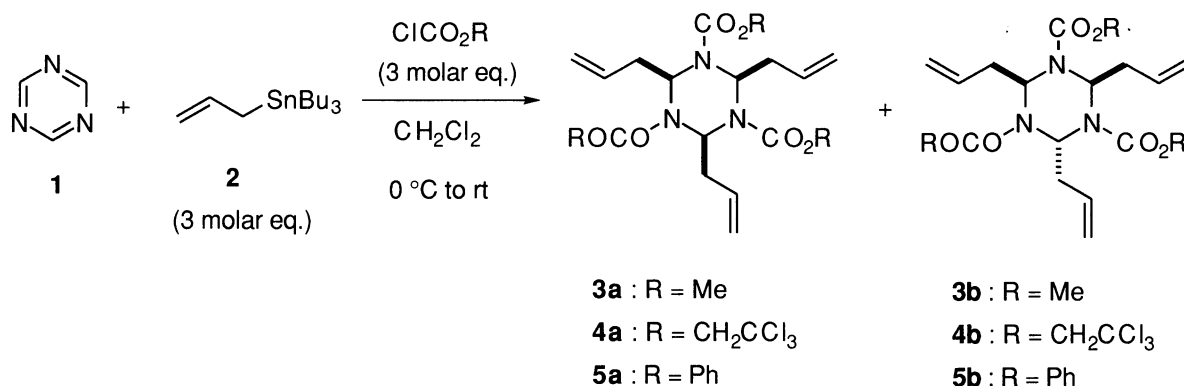


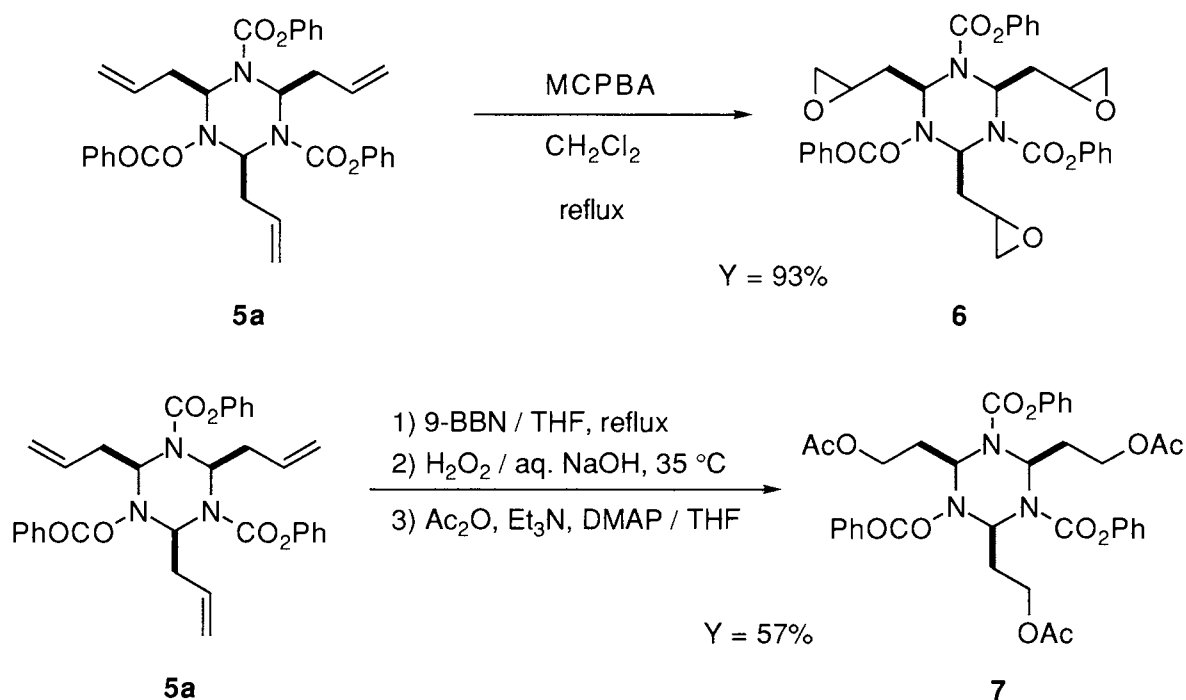
Table 1. Reaction of allyltributyltin (**2**) with 1,3,5-triazine (**1**) activated by chloroformate esters

R	Product	Yield / % <sup>a)</sup>	Ratio <sup>b)</sup>
Me	<b>3a, 3b</b>	87	<b>3a:3b</b> = 20:80
CH <sub>2</sub> CCl <sub>3</sub>	<b>4a, 4b</b>	68	<b>4a:4b</b> = 40:60
Ph	<b>5a, 5b</b>	92	<b>5a:5b</b> = 42:58

a) Isolated, combined yield. b) Determined by <sup>1</sup>H NMR.

Since syn,syn-2,4,6-triallyl-1,3,5-triazacyclohexane derivatives are geometrically interesting molecules,<sup>6)</sup> we next examined functionalization of the allyl groups. When **5a** was heated with 6 equivalents of MCPBA in CH<sub>2</sub>Cl<sub>2</sub>, epoxidation proceeded smoothly to afford syn,syn-2,4,6-tris(2,3-epoxypropyl)-1,3,5-triphenoxycarbonyl-1,3,5-triazacyclohexane (**6**) in 93% yield.<sup>7)</sup> The triepoxide **6** may be a versatile synthetic intermediate for further manipulations. In addition, **5a** was converted to the corresponding triol via hydroboration. Thus, **5a** was hydroborated with 6 equivalents of 9-BBN in THF at 60 °C. Subsequent oxidation with H<sub>2</sub>O<sub>2</sub> in aqueous NaOH at 30—35 °C followed by acetylation gave syn,syn-2,4,6-tris(3-acetoxypentyl)-1,3,5-triphenoxycarbonyl-1,3,5-triazacyclohexane (**7**) in 57% overall yield.<sup>8)</sup>

In summary, we have found a highly efficient triple allylation of 1,3,5-triazine by means of an allyltin reagent. Further manipulations of a syn,syn-triallylated product afford geometrically interesting syn,syn-2,4,6-trifunctionalized 1,3,5-triazacyclohexane derivatives.



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#### References

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- 3) All new compounds gave satisfactory analytical and spectral data.
- 4) A typical experimental procedure is as follows: To a solution of **1** (81 mg, 1.0 mmol) and **2** (1080 mg, 3.3 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (5 mL) was added  $\text{ClCO}_2\text{Ph}$  (0.42 mL, 3.0 mmol) under ice-cooling. The solution was stirred at rt overnight and the solvent was evaporated. The residue was chromatographed on silica gel (hexane /  $\text{CH}_2\text{Cl}_2$  = 10/0 to 0/10) to give **5a** (221 mg, 39%) and **5b** (304 mg, 53%). **5a**: IR (neat)  $1722, 1490\text{ cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  7.11–7.40 (m, 15H), 6.80 (br t, 3H,  $J = 8$

Hz), 5.92—6.02 (m, 3H), 5.21—5.31 (m, 6H), 3.01 (br t, 3H,  $J = 7$  Hz);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  152.4 (C), 151.0 (C), 132.9 (CH), 129.4 (CH), 125.8 (CH), 121.5 (CH), 119.0 ( $\text{CH}_2$ ), 62.1 (CH), 42.1 ( $\text{CH}_2$ ). Anal. Found: C, 69.61; H, 5.76%. Calcd for  $\text{C}_{33}\text{H}_{33}\text{N}_3\text{O}_6$ : C, 69.82; H, 5.86%. **5b**: IR (neat) 1722, 1490  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  7.14—7.44 (m, 15H), 6.45 (br t, 1H,  $J = 7$  Hz), 6.23 (br s, 2H), 5.89—6.02 (m, 3H), 5.17—5.38 (m, 6H), 3.32 (br, 2H), 2.68—2.96 (br m, 4H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  153.3 (C), 152.2 (C), 151.9 (C), 150.7 (C), 150.5 (C), 133.2 (CH), 133.0 (CH), 132.0 (CH), 129.6 (CH), 129.5 (CH), 126.1 (CH), 125.9 (CH), 121.5 (CH), 121.4 (CH), 120.4 ( $\text{CH}_2$ ), 118.7 ( $\text{CH}_2$ ), 118.5 ( $\text{CH}_2$ ), 66.0 (CH), 65.6 (CH), 65.4 (CH), 39.7 ( $\text{CH}_2$ ), 39.5 ( $\text{CH}_2$ ). Anal. Found: C, 69.86; H, 5.79%. Calcd for  $\text{C}_{33}\text{H}_{33}\text{N}_3\text{O}_6$ : C, 69.82; H, 5.86%.

- 5) All substituents of the syn,syn-stereoisomers can adopt equatorial conformation, while those of the syn,anti-stereoisomers can not.
- 6) They may be potential precursors for novel tridentate coordination compounds.
- 7) Spectral and analytical data of **6**: mp 82—83 °C; IR (KBr) 1722  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  7.17—7.42 (m, 15H), 6.97 (br, 3H), 3.19—3.31 (m, 3H), 2.83 (br s, 3H), 1.95—2.64 (br m, 9H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  152.4 (C), 152.1 (C), 151.8 (C), 151.0 (C), 150.9 (C), 150.7 (C), 129.4 (CH), 126.0 (CH), 125.9 (CH), 121.6 (CH), 121.4 (CH), 60.9 (CH), 60.8 (CH), 49.3 (CH), 49.2 (CH), 49.0 (CH), 49.0 (CH), 46.5 ( $\text{CH}_2$ ), 46.1 ( $\text{CH}_2$ ), 46.0 ( $\text{CH}_2$ ), 40.9 ( $\text{CH}_2$ ). Anal. Found: C, 64.36; H, 5.31%. Calcd for  $\text{C}_{33}\text{H}_{33}\text{N}_3\text{O}_9$ : C, 64.38; H, 5.40%.
- 8) Spectral and analytical data of **7**: IR (neat) 1733  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  7.38 (t, 6H,  $J = 7$  Hz), 7.17—7.26 (m, 9H), 6.81 (br t, 3H,  $J = 8$  Hz), 4.19 (t, 6H,  $J = 6$  Hz), 2.22—2.28 (m, 6H), 2.05 (s, 9H), 1.90—1.95 (m, 6H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  170.9 (C1), 152.5 (C), 150.8 (C), 129.5 (CH), 126.1 (CH), 121.5 (CH), 63.4 ( $\text{CH}_2$ ), 62.3 (CH), 33.9 ( $\text{CH}_2$ ), 25.5 ( $\text{CH}_2$ ), 20.9 ( $\text{CH}_3$ ). Anal. Found: C, 62.43; H, 5.97%. Calcd for  $\text{C}_{39}\text{H}_{45}\text{N}_3\text{O}_{12}$ : C, 62.64; H, 6.06%.

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